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Plan for Solving Boyle's Law Problems Rearrange the equation  $P_1V_1 = P_2V_2$

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## Skills Worksheet Problem Solving

Holt ChemFile: Problem-Solving Workbook 3 Conversions Name Class Date Problem Solving continued The correct conversion factor is the one that when multiplied by the given quantity causes the units to cancel. COMPUTE 45.5 g calcium chloride  $100 \frac{1 \text{ g}}{10 \text{ mg}}$  45 500 mg calcium chloride EVALUATE Are the units correct? Yes; milligrams are the desired units.

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Holt ChemFile: Problem-Solving Workbook 99 Stoichiometry Name Class Date Problem Solving continued Sample Problem 1 Ammonia is made industrially by reacting nitrogen and hydrogen under pressure, at high temperature, and in the presence of a catalyst. The equation is  $\text{N}_2(\text{g}) + 3\text{H}_2(\text{g}) \rightarrow 2\text{NH}_3(\text{g})$ . If 4.0 mol of  $\text{H}_2$  react, how many moles of  $\text{NH}_3$  will be produced?

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Name Class Date Problem Solving continued Practice 1. Complete the  
table below using the following equation, which represents a reaction  
that produces aluminum chloride.  $2\text{Al}(\text{s}) + 3\text{Cl}_2(\text{g}) \rightarrow 2\text{AlCl}_3(\text{s})$   
Mass Al Volume  $\text{Cl}_2$  Conditions Mass  $\text{AlCl}_3$  a. excess ?

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Solving continued Is the answer reasonable? Yes; 2 g of boron is about  
1/5 of the molar mass of boron. Therefore, 2.00 g boron will contain  
...

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